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P.G. Diploma in Environmental Monitoring and Impact Assessment Annual Examination-2010 Paper No: PGDEMIA-103

Instrument in Environment Monitoring

Time: 2 /2 hours

Maximum marks: 70

Attempt all questions from the section I, any six questions from section II and any three questions from section III

Section I

1. pH indicators change its color at certain

1×10=10

- a) pH
- b) Temperature
- c) Humidity
- d) None of the above
- 2. The addition of acids to water causes the concentration of hydrogen ions to be
 - a) Greater than 1 × 10⁻⁷ mole/l
 - b) Lesser than 1 x 10⁻⁷ mole/l
 - c) Greater than 1 x 10 mole/l
 - d) Lesser than 1 x 10 mole/l
- 3. Litmus is extracted from
 - a) Bacteria
 - b) Fungus
 - c) Lichens
 - d) None of the above
- 4. On increasing temperature the conductivity of a solution is
 - a) Increased
 - b) Decreased
 - c) Stable
 - d) None of the above
- 5. Low speed centrifuge typically attain rotation speeds
 - a) More than 10,000 rpm
 - b) Less than 10,000 rpm
 - c) More than 30,000 rpm
 - d) More than 1,20,000 rpm

Section-II

- Enlist the devices used for measuring volume and discuss pipettes in detail 6x5= 30
- 2. Write informative notes on any two
 - a) Basic design of conductivity meter
 - b) Basic principle of centrifuge
 - c) Types of balances
 - d) Filtration systems
- Explain principle of gas chromatography. Difference between packed column and opened tubular column
- 4. Write a note on classification of chromatography
- 5. Explain principle and application of UV visible spectrophotometer
- 6. Draw flow diagram of flame photometer and atomic absorption spectrometer.
- 7. Discuss supporting media used in the electrophoresis
- 8. Explain principle and application of electrophoresis

Section III

Give the detailed account of pH indicators and pH meter.

10x3=30

- 2. Explain principle and construction of high performance liquid chromatography
- 3. Write basic principle, application and instrumentation of TLC
- 4. Explain principle and components of atomic absorption spectroscopy
- Discuss high volume air samples